



Supplement of

**When colour makes minerals unique: the case of
the green dolomite from Malentrata (Pomarance,
Tuscany, Italy)**

Luca Adami et al.

Correspondence to: Luca Adami (luca.adami@unife.it)

The copyright of individual parts of the supplement might differ from the article licence.

Figures

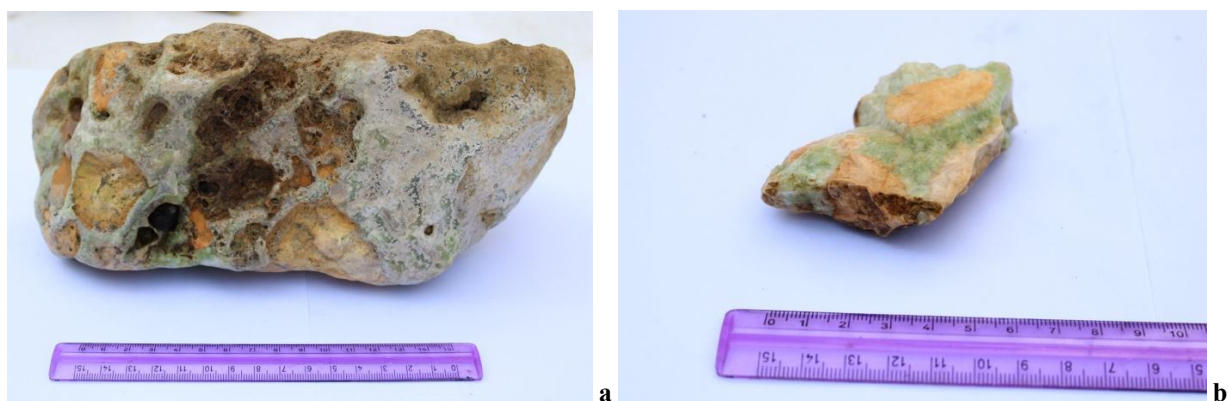


Figure S1a and S1b – Dolomite-magnesite samples from the Malentrata mine.

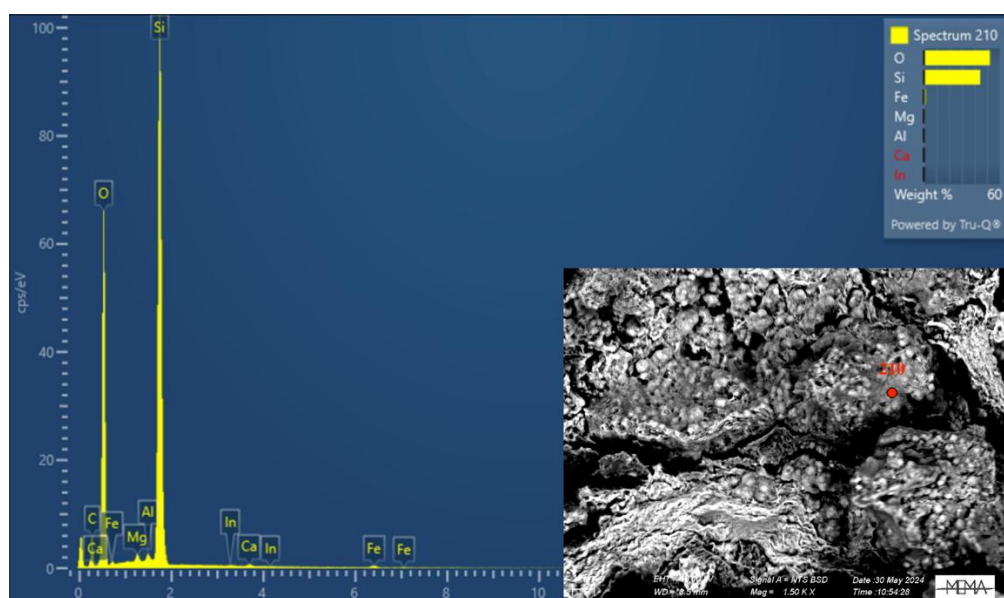


Figure S2 – EDS spectrum of part of the dolomite sample from Figure 3b. The chemical composition and habit correspond to those of an alteration product (e.g. a phyllosilicate).

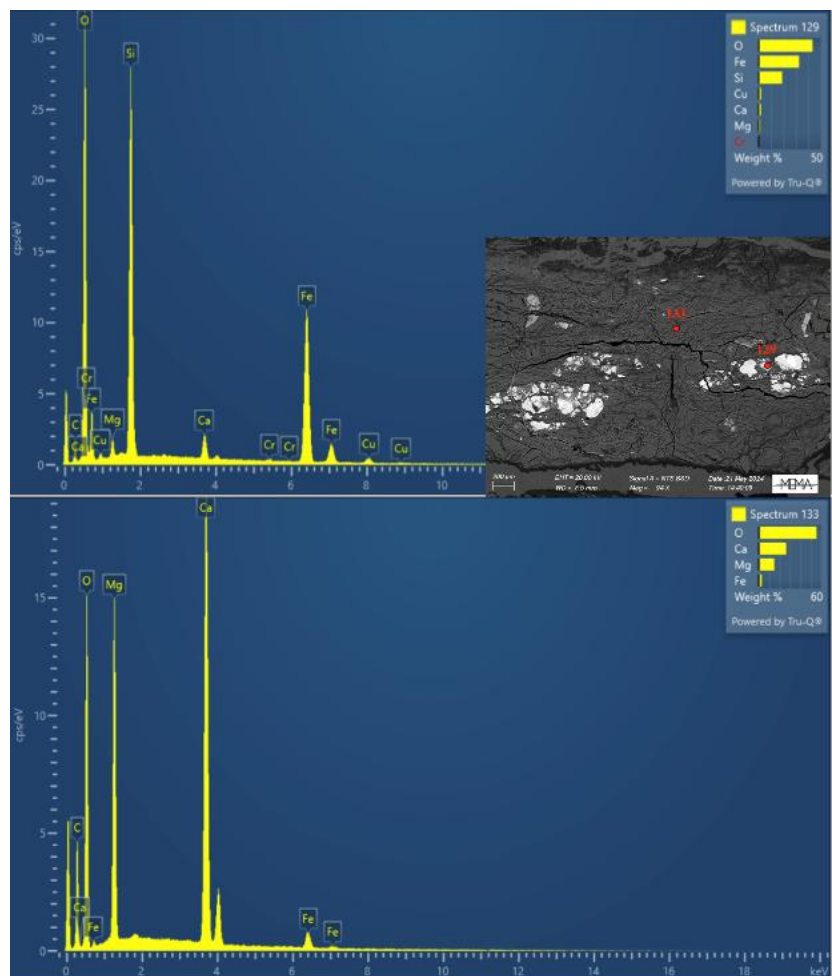


Figure S3 – EDS spectra of part of the dolomite sample from Figure 3c. The chemical composition is that of a magnesite (spectrum 133) and mineralised phases containing iron and silicon (spectrum 129).

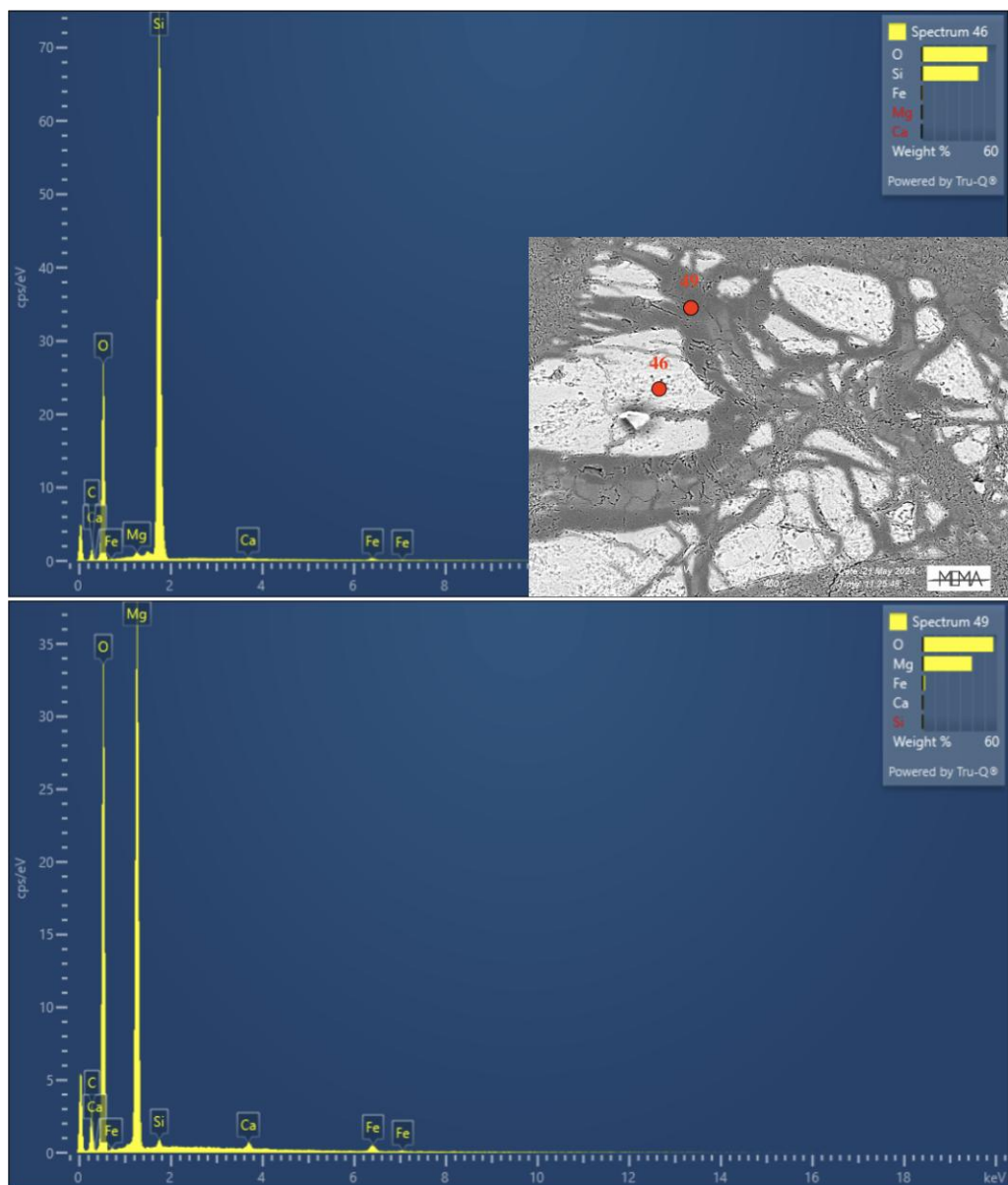


Figure S4 – EDS spectra of part of the dolomite sample from Figure 3d, where a fragment of chalcedony (spectrum 46) is surrounded by veins of magnesite (spectrum 49).

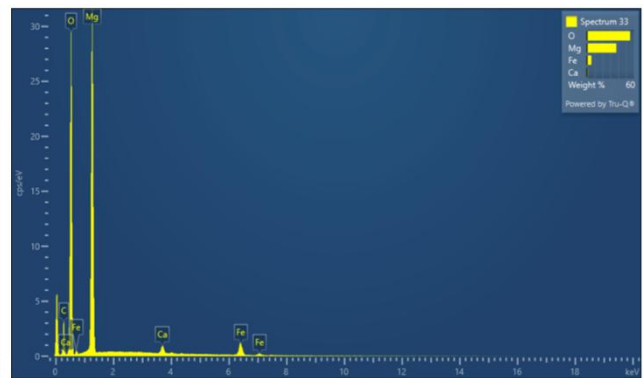
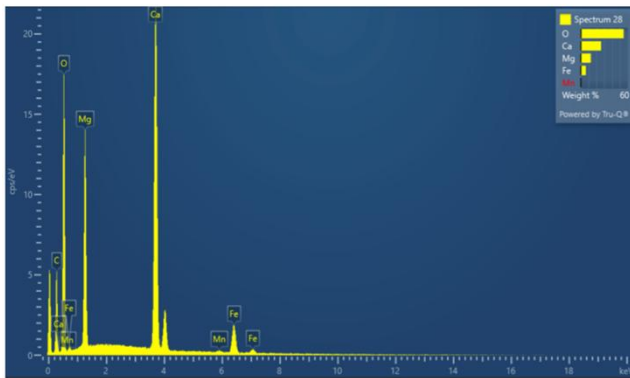
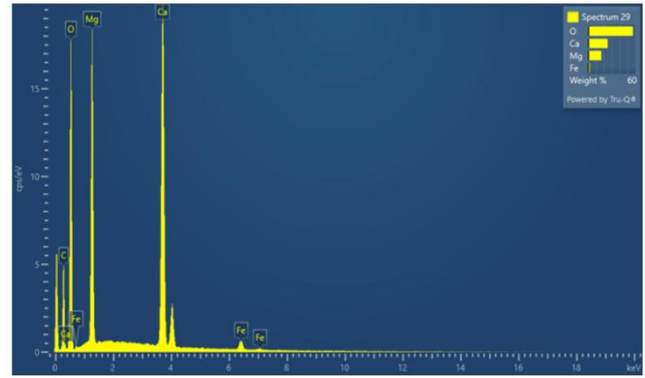
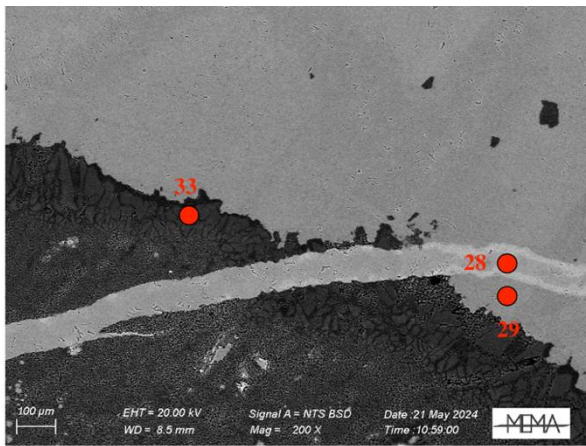


Figure S5 – EDS spectra of a part of the dolomite sample from Figure 3e. The late dolomite vein (spectrum 28) cuts through both early dolomite (spectrum 29) and magnesite (spectrum 33).

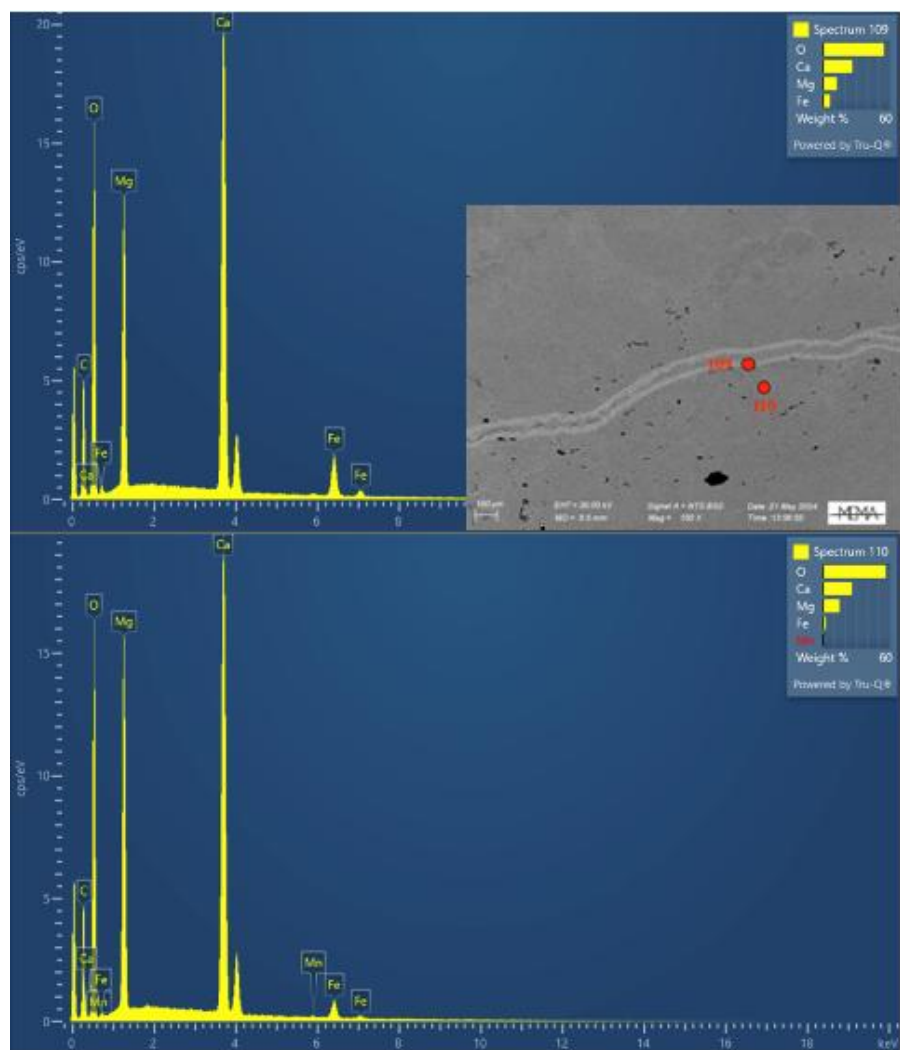


Figure S6 – EDS spectra of a part of the dolomite sample from Figure 3f. Spectra 109 and 110 are associated with a dolomitic composition. The edges of the dolomite vein show a higher concentration of iron.

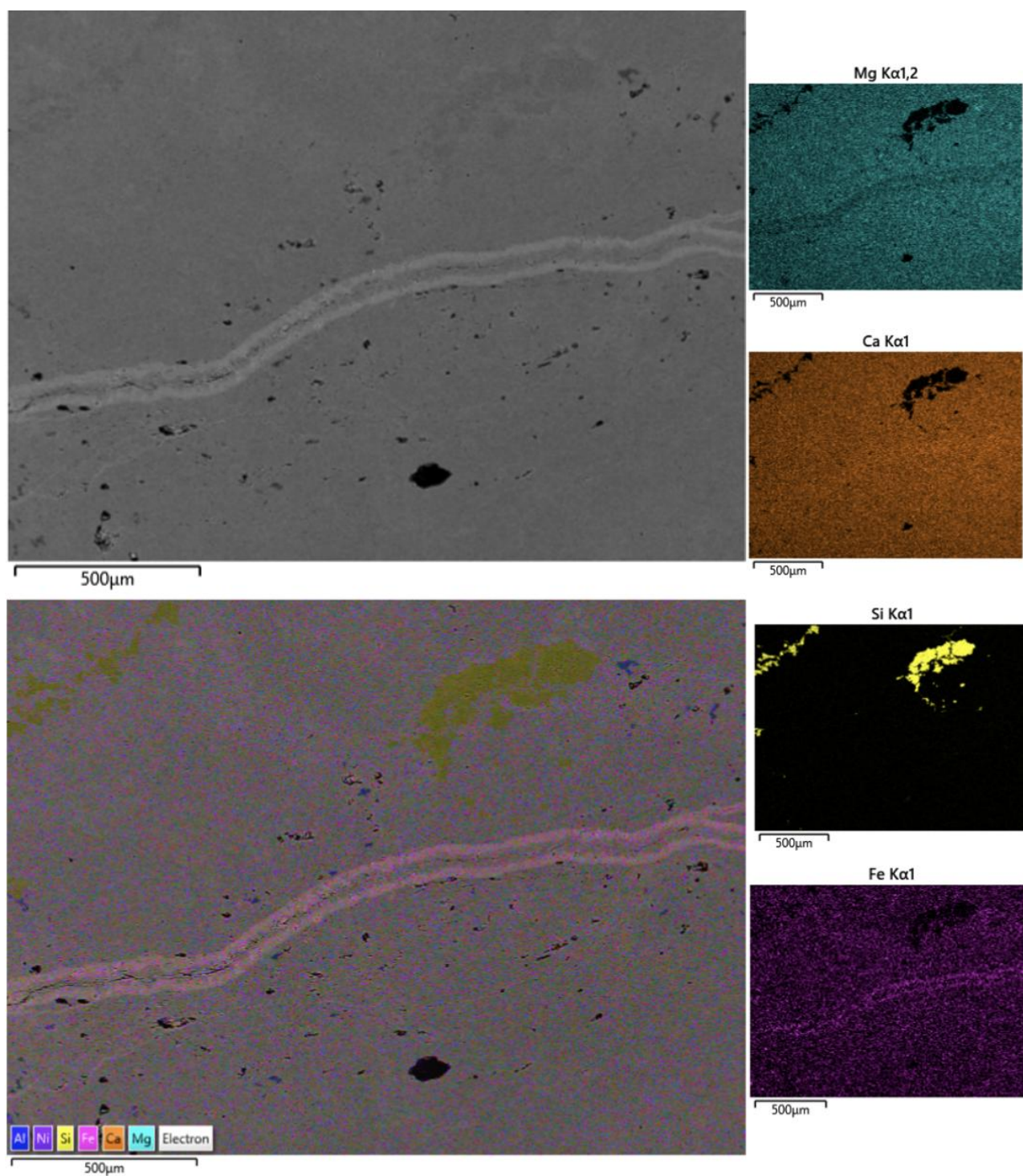


Figure S7 – Chemical mapping of part of the dolomite sample from Figure 3f.

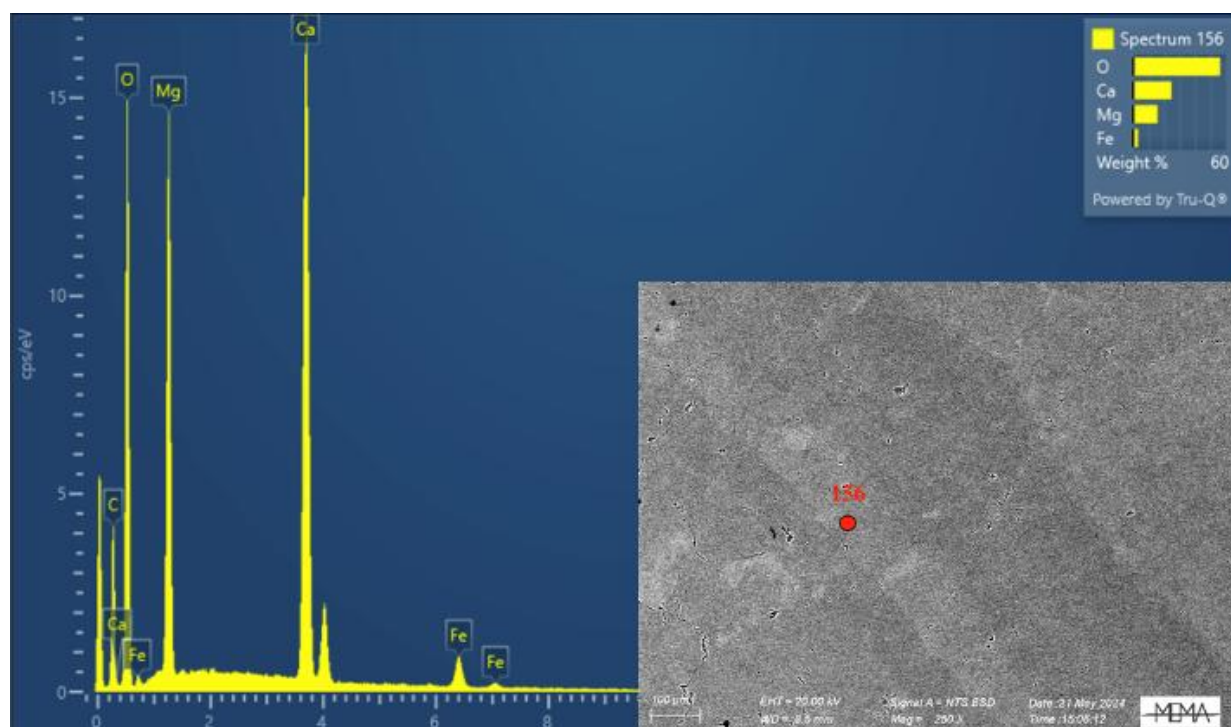


Figure S8 – EDS spectrum of part of the dolomite sample from Figure 3g. The micrograph and corresponding EDS spectrum show the zoning of the dolomite.

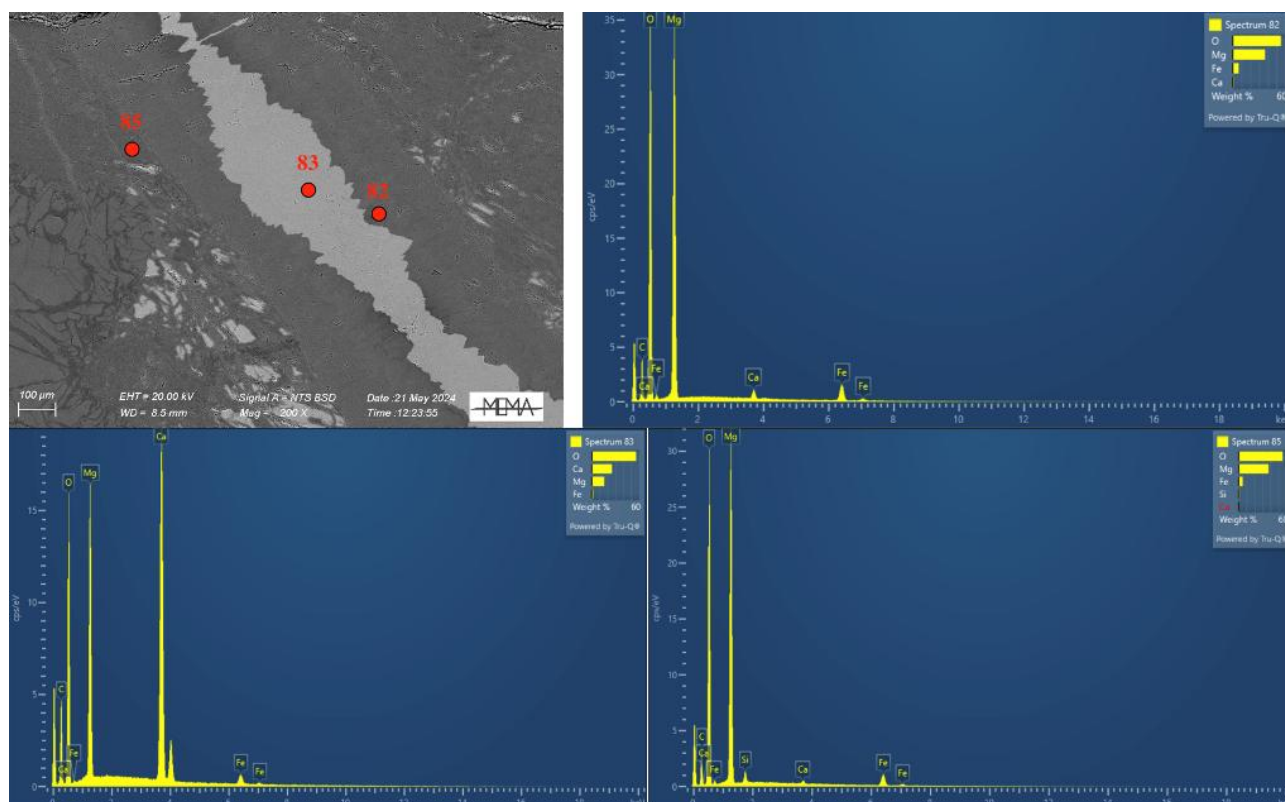


Figure S9 – EDS spectrum of part of the dolomite sample from Figure 3h. A dolomite vein (spectrum 83) intersects a darker area of magnesite composition (spectra 82 and 85).

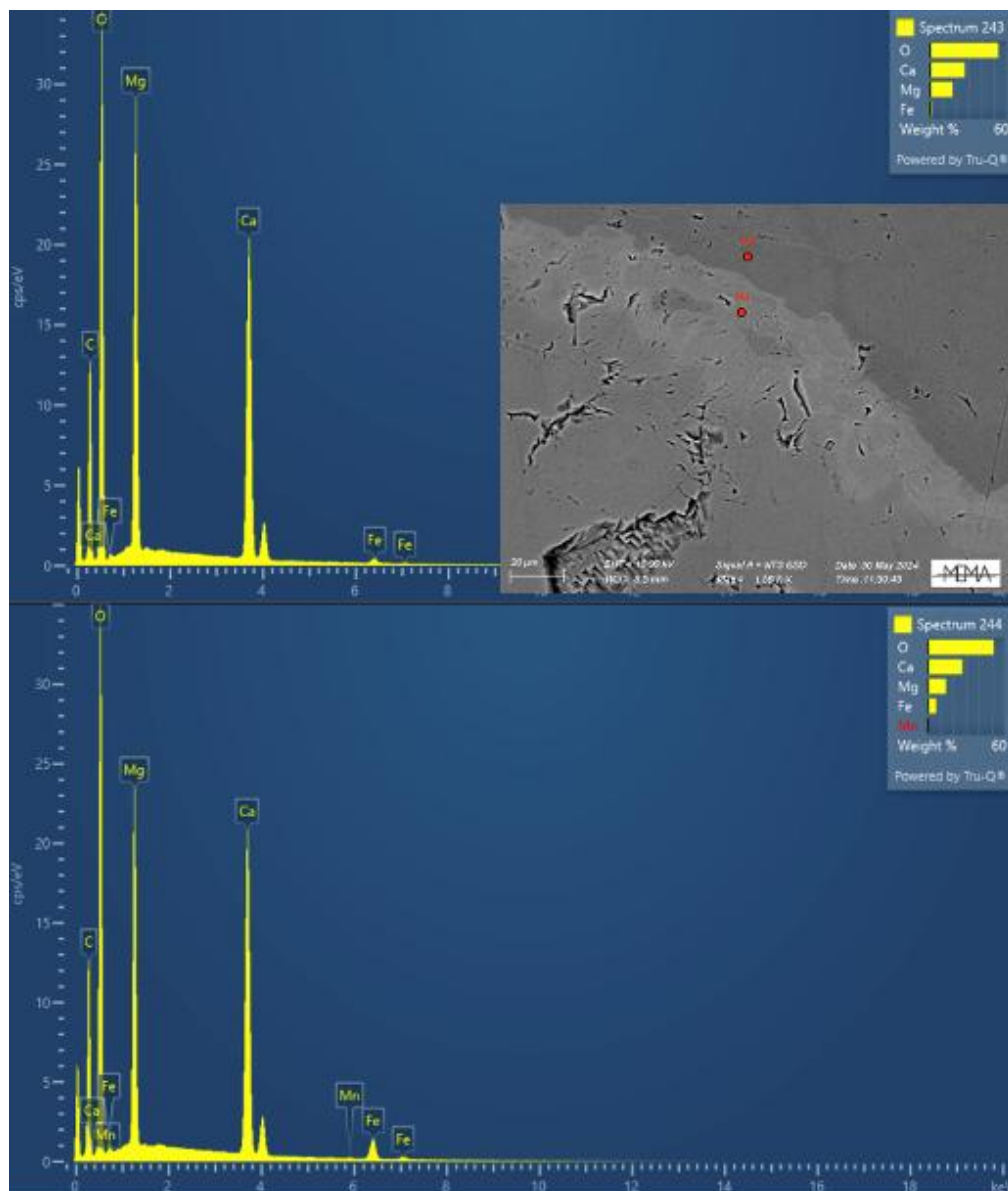


Figure S10 – EDS spectrum of part of the dolomite sample from Figure 3i. The BSE image represents two magnesite compositions with different iron concentrations. Specifically, spectrum 243 represents a magnesite with a lower iron concentration than spectrum 244.

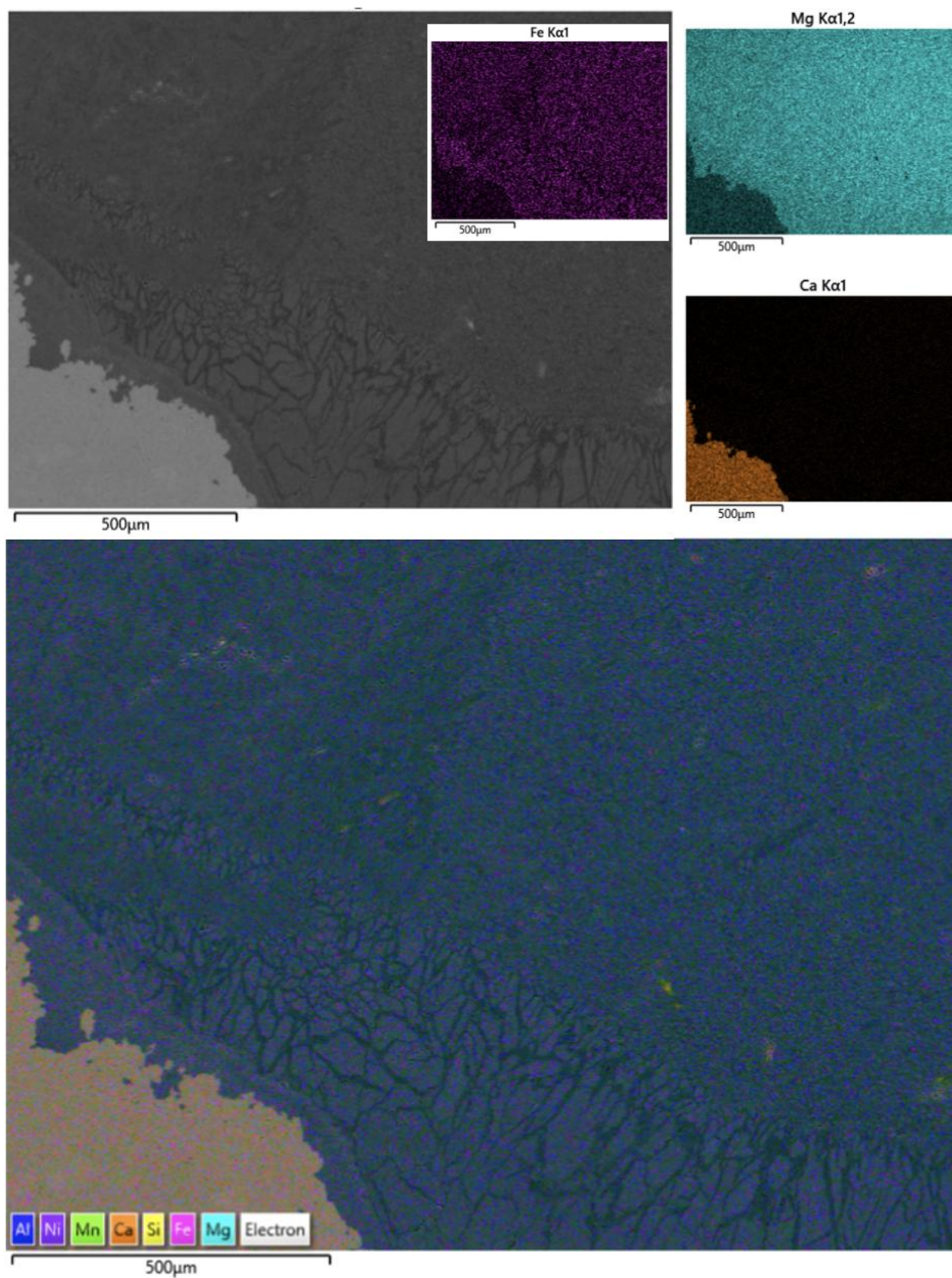


Figure S11 – Chemical mapping of a boundary region between dolomite and magnesite.

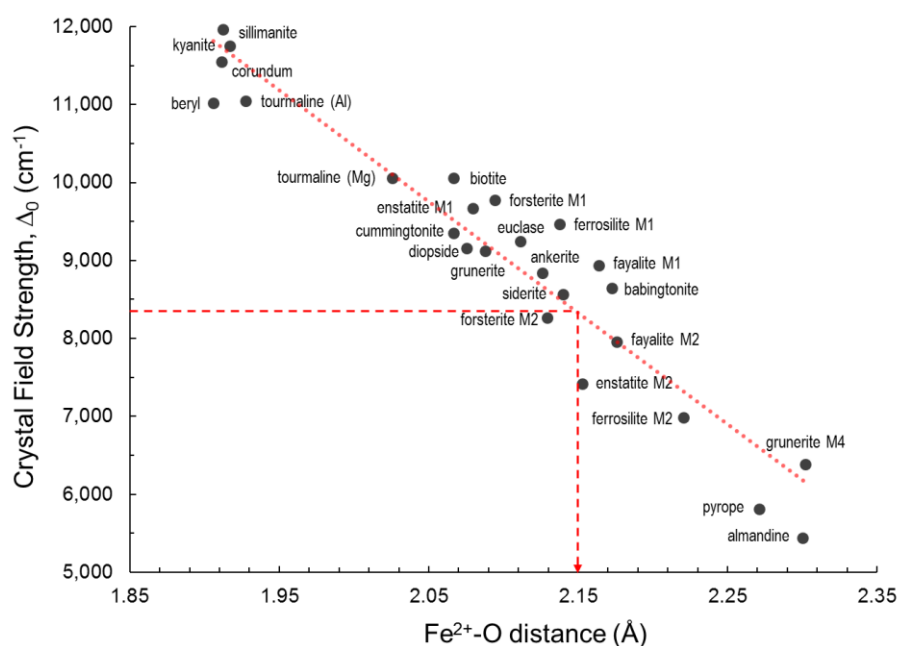


Figure S12- Fe^{2+} -O distance in silicates and carbonates as a function of Crystal Field Strength (data from (Burns, 1993; Lavina et al., 2010; Ross and Reeder, 1992; Taran et al., 2017)). Dashed line: range of possible Fe^{2+} -O distance for dolomite from Malenratra Mine.

Tables

Table S1 – RT ^{57}Fe hyperfine parameters obtained for the sample analysed by Mössbauer analysis. δ is quoted relative to metallic iron at room temperature; the error on the relative areas is estimated to be 5%.

	δ (mm/s)	Δ (mm/s)	Γ_+ (mm/s)	A (%)	Fe(III)/Fe(II)	Attribution
Doublet 1	1.258 ± 0.004	1.743 ± 0.004	0.192 ± 0.005	98	0.02	Fe(II) M
Doublet 2	0.35*	0.55*	0.2*	2*		Fe(III) M

References

Inorganic crystal structure (ICSD) database number and reference for the unit-cell parameters of dolomite at ambient conditions.

- 10404: Althoff, P. L. (1977). Structural refinements of dolomite and a magnesian calcite and implications for dolomite formation in the marine environment. *American Mineralogist*, **62**(7-8), 772–783.
- 31209: Reeder, R. J., & Wenk, H. R. (1983). Structure refinements of some thermally disordered dolomites. *American Mineralogist*, **68**(7-8), 769–776.

- 31276: Kirfel, A., & Will, G. (1983). Untersuchungen zur Elektronendichteverteilung im Dolomit, $\text{CaMg}(\text{CO}_3)_2$. *Tschermaks mineralogische und petrographische Mitteilungen*, **31**, 151–164. Zobetz, E, 60–68.
- 40968, 202162: Miser, D. E., Swinnea, J. S., & Steinfink, H. (1987). TEM observations and X-ray crystal-structure refinement of a twinned dolomite with a modulated microstructure. *American Mineralogist*, **72**(1-2), 188–193.
- 52149: Pilati, T., Demartin, F., & Gramaccioli, C. M. (1998). Lattice-dynamical estimation of atomic displacement parameters in carbonates: calcite and aragonite CaCO_3 , dolomite $\text{CaMg}(\text{CO}_3)_2$ and magnesite MgCO_3 . *Acta Crystallographica Section B: Structural Science*, **54**(5), 515–523.
- 66333: Ross, N. L., & Reeder, R. J. (1992). High-pressure structural study of dolomite and ankerite. *American Mineralogist*, **77**(3-4), 412–421.
- 87088: Suzuki, Y., Morgan, P. E., & Niihara, K. (1998). Use of a high X-ray flux instrument for a mineral: X-ray powder diffraction pattern of $\text{CaMg}(\text{CO}_3)_2$. *Powder Diffraction*, **13**(4), 216–221.
- 100416: Beran, A., & Zemann, J. (1977). Refinement and comparison of the crystal structures of a dolomite and of an Fe-rich ankerite. *Mineralogy and Petrology*, **24**(4), 279–286.
- 100680: Effenberger, H., Mereiter, K., & Zemann, J. (1981). Crystal structure refinements of magnesite, calcite, rhodochrosite, siderite, smithonite, and dolomite, with discussion of some aspects of the stereochemistry of calcite type carbonates. *Zeitschrift für Kristallographie-Crystalline Materials*, **156**(1-4), 233–244.
- 152201: Drits, V. A., McCarty, D. K., Sakharov, B., & Milliken, K. L. (2005). New insight into structural and compositional variability in some ancient excess-Ca dolomite. *Canadian mineralogist*, **43**(4), 1255.
- 171508: Antao, S. M., Mulder, W. H., Hassan, I., Crichton, W. A., & Parise, J. B. (2004). Cation disorder in dolomite, $\text{CaMg}(\text{CO}_3)_2$, and its influence on the aragonite + magnesite $\leftrightarrow$ dolomite reaction boundary. *American Mineralogist*, **89**(7), 1142–1147.
- 185042: Zucchini, A., Comodi, P., Katerinopoulou, A., Balic-Zunic, T., McCammon, C., & Frondini, F. (2012). Order–disorder–reorder process in thermally treated dolomite samples: a combined powder and single-crystal X-ray diffraction study. *Physics and Chemistry of Minerals*, **39**, 319–328.
- 188932: Sitepu, H., & Al-Ghamdi, R. A. (2013). Use of the Rietveld Method for Describing Structure and Texture in XRD Data of Dolomite $[\text{CaMg}(\text{CO}_3)_2]$ and Hydromagnesite $[\text{Mg}_5(\text{CO}_3)_4(\text{OH})_2(\text{H}_2\text{O})_4]$ Powders. *Advances in X-ray Analysis*, **56**, 137–145.
- 193066: Zucchini, A., Comodi, P., Nazzareni, S., & Hanfland, M. (2014). The effect of cation ordering and temperature on the high-pressure behaviour of dolomite. *Physics and Chemistry of Minerals*, **41**, 783–793.